The Antimicrobial Spectrum of Disinfectants

This table provides general information for selected disinfectant chemical classes. Antimicrobial activity may vary with formulation and concentration. The use of trade names does not in any way signify endorsement of a particular product. They are provided as examples.

Removal of organic material Peroxygens must always precede the use accelerated Phenolic Quaternary hydrogen peroxide Alkalis **Biguanides** Compounds Ammonium of any disinfectant. **Acids Aldehvdes** sodium hydroxide. chlorhexidine. (Rescue®), potassium (Lvsol®, Osvl®, Compounds Halogens **Alcohols** peroxymonosulfate hydrochloric formaldehyde, ammonium Nolvasan®, Amphyl[®], (Roccal", Zepharin", sodium (Virkon-S[®]), peroxyacetic acid, acetic acid, ethanol, paraformaldehyde, hydroxide, sodium ChlorHex®, TekTrol[®], DiQuat", Parvosol", most susceptible hypochlorite iodine acid, (Oxy-Sept 333) Pheno-Tek II®) D-256°) citric acid isopropanol gluteraldehyde carbonate Virosan® mycoplasmas ++ ++ ++ ++ ++ ++ ++ ++ ++ gram-positive bacteria ++ ++ ++ ++ ++ ++ ++ gram-negative bacteria ± ++ ++ ++ pseudomonads susceptibility of microorganisms to chemical disinfectants ± rickettsiae ± **+** a ++ enveloped viruses ± chlamydiae non-enveloped viruses ± ± fungal spores Ν picornaviruses (i.e. FMD) Ν parvoviruses acid-fast bacteria bacterial spores + c coccidia prions a-varies with composition highly effective LEGEND the Center for Food Security & Public Health b-peracetic acid is sporicidal no activity most resistant c-ammonium hydroxide effective N information not available d-some have activity against coccidia limited activity

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